

Thermal Physics II

1st short test – 26 February 2013

given and surname : Lecturer

university number :

course of study : Teaching

marks obtained : 26

total marks : 25 + 1

comments :

1. Order the following thermodynamic quantities into a group of extensive quantities on the left and a group of intensive quantities on the right: internal energy U , energy density $u = U/V$, volume V and entropy S .

$$U, V, S \qquad u = U/V$$

2 $\frac{1}{2}$ each

2. State the 1st law of thermodynamics for a gas of atoms.

$$dU = \delta Q + \delta W = \delta Q - p dV$$

2

3. A small amount of heat, ΔQ , is transferred to a system at temperature T via a quasi-static process. What is the resulting change in entropy?

$$\Delta S = \frac{1}{T} \Delta Q$$

2

4. How does the heat capacity of a solid at constant magnetic field behave when the system is cooled close to $T = 0$?

$$\lim_{T \rightarrow 0} C_B = 0$$

1

5. Which of the following quantities has an exact differential in the variables stated: $U = U(S, V)$, $U = U(T, V)$, $S = S(U, V)$, $Q = Q(S, T)$?

$$U = U(S, V) \quad \text{and} \quad S = S(U, V)$$

2

6. State the relation that defines the temperature for a isolated system.

$$T = \left(\frac{\partial U}{\partial S} \right)_V$$

1

7. A system has 5 microstates with energies $E_1 = 2$, $E_2 = 3$, $E_3 = 4$, $E_4 = 5$ and $E_5 = 1$. Calculate the average energy and its fluctuation σ_E .

$$\begin{aligned} \langle E \rangle &= \frac{1}{N} \sum E_i & \sigma_E^2 &= \langle (E_i - \langle E \rangle)^2 \rangle \\ &= 3 & &= \frac{1}{5} [1+0+1+4+4] \\ & & &= 2 \quad \sigma_E = \sqrt{2} \end{aligned}$$

4

8. At different times, the entropy of an isolated systems has been measured to be i) $S = 3 \text{ J/K}$, ii) $S = 1.5 \text{ J/K}$, iii) $S = 3.5 \text{ J/K}$, iv) $S = 2.5 \text{ J/K}$. Which of the states probed is closest to equilibrium?

$$\text{case iii) } S = 3.5 \text{ J/K}$$

1

9. A thermodynamic system is composed of two independent parts having entropies S_1 and S_2 and number of states Ω_1 and Ω_2 , respectively. What is the entropy and the number of states for the combined system?

$$S_{\text{tot}} = S_1 + S_2 \quad \Omega_{\text{tot}} = \Omega_1 \cdot \Omega_2$$

2

10. Consider an idealised coin tossing experiment (same probabilities for heads and tails). What is the *probability* to get exactly 2 heads in 5 tries?

$$p = \frac{w_5(2)}{\Omega} \quad w_5(2) = \frac{5!}{2!3!} = 10$$

$$= \frac{10}{2^5} = \frac{5}{16} \quad \Omega = 2^5$$

4

11. What is the number of states for a system with 20 entirely independent particles that can each occupy 5 states independent of each other?

$$5^{20}$$

1

12. Using the fact that $dU = TdS - pdV + \mu dN$, derive the Maxwell-relation $(\partial T/\partial N)_{S,V} = (\partial \mu/\partial S)_{V,N}$.

$$T = \left(\frac{\partial U}{\partial S}\right)_{V,N} \quad -p = \left(\frac{\partial U}{\partial V}\right)_{S,N} \quad \mu = \left(\frac{\partial U}{\partial N}\right)_{S,V}$$

$$\left(\frac{\partial}{\partial N} \left(\frac{\partial U}{\partial S}\right)_{V,N}\right)_{S,V} = \left(\frac{\partial}{\partial S} \left(\frac{\partial U}{\partial N}\right)_{S,V}\right)_{V,N}$$

$$\Rightarrow \left(\frac{\partial T}{\partial N}\right)_{S,V} = \left(\frac{\partial \mu}{\partial S}\right)_{V,N}$$

3

A+ A family has two children. You are told that one of them is a boy. What is the probability that the family has two boys?

$$\frac{1}{3}$$

1 extra