Thermal Physics II

$2^{\rm nd}$ short test - 13 March 2013

given and surname :

university number :

course of study :

marks obtained :

total marks :

comments :

1.	How are the Helmholtz free energy and the internal energy related?
2.	What are the natural variables that allow for the use of the Helmholtz free energy as a thermodynamic potential for a gas?
3.	State the relation between the number of states of the system and the number of states of its heat reservoir within the canonical description.
4.	Define the Boltzmann factor for a quantum state with energy E_1 and the partition functions of a system with discrete energies E_j .
5.	Consider an ion with two energy states at $E_1 = -30 \mathrm{eV}$ and $E_2 = -20 \mathrm{eV}$ embedded in a gas with a temperature such that $k_B T = 10 \mathrm{eV}$. Calculate the ratio of the occupation probabilities for these two states.
6.	Give the basic relation that connects the thermodynamical and statistical descriptions of many-body systems in the canonical ensemble.

8.	A system is made of two independent subsystems with internal energies U_1 and U_2 , and partition functions, Z_1 and Z_2 . State the internal energy partition function and Helmholtz free energy of the combined system.
9.	Which quantity in the statistical description of many-body systems is defined by the temperature of the system? What happens to this quantit if the temperature is doubled?
0.	State the partition function for an ideal gas $(N \text{ particles in volume } V)$.
1.	Given the fact that the differential of the Gibbs energy (enthalpy) is given by $dH = TdS + Vdp$, derive the Maxwell-relation $(\partial T/\partial p)_S = (\partial V/\partial S)_p$